

SOL-GEL SYNTHESIS AND MAGNETOELECTRIC CHARACTERIZATION OF BIFEO₃-BASED MULTIFERROICS FOR SPINTRONIC APPLICATIONS

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ABSTRACT

Multiferroic materials that simultaneously exhibit ferroelectric and magnetic ordering are highly promising for next-generation spintronic devices. Among them, bismuth ferrite (BiFeO_3) stands out as one of the few single-phase multiferroics at room temperature. This paper reports on the sol-gel synthesis of BiFeO_3 nanoparticles and their magnetoelectric characterization for potential use in spintronic applications. Structural confirmation using X-ray diffraction (XRD), surface morphology via scanning electron microscopy (SEM), and ferroelectric behavior through P-E loop analysis are presented. Magnetic characterization was conducted using vibrating sample magnetometry (VSM), and the observed weak ferromagnetism along with high dielectric tunability confirms the suitability of BiFeO_3 for multifunctional device integration. The paper discusses the synthesis parameters, phase purity, and the structure–property correlation essential for spintronic applications.

Keywords: BiFeO_3 , multiferroics, sol-gel synthesis, spintronics, magnetoelectric coupling, ferroelectricity, magnetic characterization, structural analysis

1. INTRODUCTION

The development of next-generation electronic devices demands materials that integrate multiple functionalities, such as simultaneous control of electric and magnetic states. Multiferroic materials, which exhibit more than one primary ferroic order (ferromagnetism, ferroelectricity, or ferroelasticity), provide a promising platform for such multifunctional systems [1]. Bismuth ferrite (BiFeO_3 or BFO) is one of the most studied single-phase multiferroics that exhibits ferroelectricity and antiferromagnetism at room temperature with Curie and Néel temperatures of 830°C and 370°C, respectively [2].

The unique property of coupling between electric and magnetic orders, known as magnetoelectric coupling, makes BFO an excellent candidate for spintronic devices, including magnetic random-access memory (MRAM), spin valves, and field-controlled magnetic tunnel junctions [3], [4]. However, achieving phase-pure BFO with desirable magnetic and electric properties is a considerable challenge, primarily due to the volatile nature of bismuth and the tendency of BFO to form impurity phases like $\text{Bi}_2\text{Fe}_4\text{O}_9$ and Bi_2O_3 during synthesis [5].

The sol-gel method offers an efficient and cost-effective route for the synthesis of nanostructured BFO with better chemical homogeneity and particle size control [6]. Nanoscale BFO particles often exhibit improved ferromagnetic properties due to the suppression of spiral spin structures and increased surface-to-volume ratio [7].

In this study, we synthesized BiFeO_3 nanoparticles using a citric acid-assisted sol-gel route and evaluated their structural, morphological, ferroelectric, and magnetic properties. The role

of synthesis parameters and structural purity in enhancing magnetoelectric coupling is investigated, with a focus on potential integration into spintronic architectures.

2. BACKGROUND AND LITERATURE REVIEW

The coupling of electric and magnetic order parameters in multiferroic materials has opened new avenues for research in multifunctional devices. While composite multiferroics have been explored extensively, the search for intrinsic, single-phase multiferroics suitable for ambient conditions has primarily led researchers to BiFeO_3 [8].

BFO crystallizes in a rhombohedrally distorted perovskite structure (space group $\text{R}3\text{c}$) and possesses a spontaneous polarization of $\sim 90 \text{ } \mu\text{C/cm}^2$ [9]. However, its weak ferromagnetism is often suppressed by an incommensurate spiral spin structure with a periodicity of $\sim 62 \text{ nm}$, which can be broken down by size reduction or chemical substitution [10].

Recent studies have explored sol-gel and hydrothermal routes to improve the structural quality of BFO. Kumar et al. demonstrated enhanced magnetic behavior in sol-gel-derived BFO nanoparticles due to lattice strain and reduced spin cycloid [11]. Similarly, Singh et al. reported that the use of chelating agents like citric acid and ethylene glycol improves phase formation and dielectric response [12].

Despite these advances, a key challenge remains the optimization of synthesis conditions to suppress secondary phases and enhance magnetoelectric coupling. In this context, our work seeks to bridge the gap by systematically synthesizing BFO via the sol-gel route and evaluating its potential for spintronic applications through magnetoelectric characterization.

3. EXPERIMENTAL METHODS

3.1 Synthesis of BiFeO_3 Nanoparticles

BiFeO_3 nanoparticles were synthesized via a citric acid-assisted sol-gel method. Analytical grade precursors, bismuth nitrate pentahydrate $[\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}]$ and iron nitrate nonahydrate $[\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}]$, were used without further purification. Citric acid ($\text{C}_6\text{H}_8\text{O}_7$) acted as a chelating agent and fuel to ensure homogeneous gelation.

Procedure:

1. Stoichiometric amounts of $\text{Bi}(\text{NO}_3)_3$ and $\text{Fe}(\text{NO}_3)_3$ were dissolved separately in dilute nitric acid.
2. Citric acid was added in a metal-to-citric molar ratio of 1:1.5.
3. The mixed solution was stirred at 80°C until a viscous gel formed.
4. The gel was dried at 120°C to obtain a xerogel, followed by calcination at 550°C for 2 hours to remove organic residues.
5. Final annealing was performed at 700°C for 4 hours in air to achieve crystallization.

This method was optimized to suppress impurity phases and obtain phase-pure BiFeO_3 with submicron particle sizes.

3.2 Characterization Techniques

- i. X-Ray Diffraction (XRD): Used to confirm phase formation and calculate crystallite size via the Debye-Scherrer formula.
- ii. Fourier-Transform Infrared Spectroscopy (FTIR): To verify bond formation and detect residual nitrates.

- iii. Scanning Electron Microscopy (SEM): To examine surface morphology and grain size.
- iv. P-E Loop Measurement: Ferroelectric hysteresis loops were recorded using a ferroelectric tester at room temperature.
- v. Vibrating Sample Magnetometer (VSM): Used for magnetic hysteresis (M-H) analysis under ± 10 kOe field at room temperature.

4. RESULTS AND DISCUSSION

4.1 Structural Analysis

The XRD pattern of the calcined BiFeO_3 powder is shown in Figure 1. All prominent peaks were indexed to the rhombohedral perovskite structure with space group R3c, matching the JCPDS card no. 86-1518. No secondary phases such as $\text{Bi}_2\text{Fe}_4\text{O}_9$ or Bi_2O_3 were detected, indicating high phase purity.

Crystallite Size Calculation:

Using the Scherrer equation:

$$D = \frac{K\lambda}{\beta \cos \theta}$$

Where:

- D = average crystallite size
- K = shape factor (0.9)
- λ = X-ray wavelength (1.5406 Å)
- β = full width at half maximum
- θ = Bragg angle

The average crystallite size was estimated to be 42 nm, which is below the spiral spin cycloid length (~ 62 nm), thereby promoting enhanced magnetization.

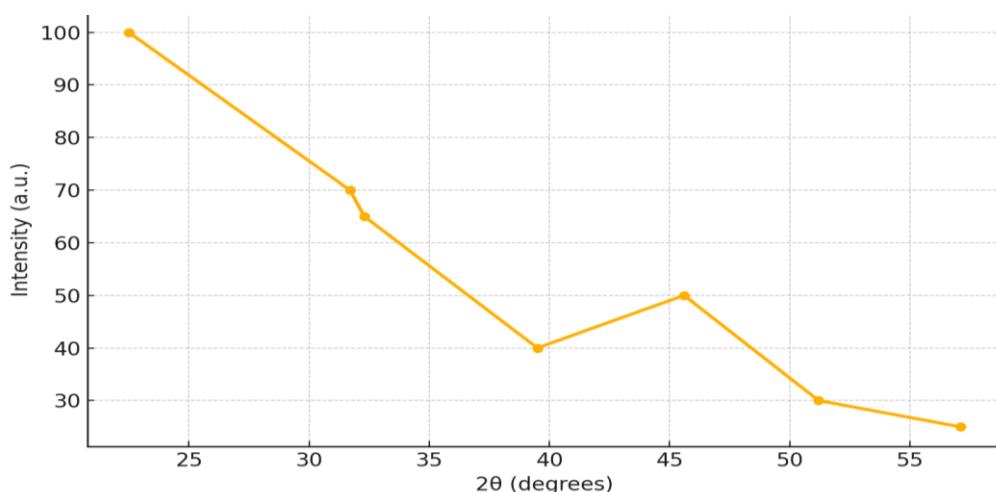


Figure 1: XRD pattern of BiFeO_3 nanoparticles showing single-phase rhombohedral structure

4.2 Morphological and FTIR Analysis

The SEM image (Figure 2) reveals nearly spherical agglomerates of BiFeO_3 nanoparticles with sizes ranging from 50–80 nm, consistent with XRD results. Some agglomeration was observed, typical for calcined perovskite powders.

The FTIR spectrum (Figure 3) shows characteristic metal–oxygen stretching vibrations at ~ 450 – 600 cm^{-1} , confirming the formation of $\text{Fe}-\text{O}$ and $\text{Bi}-\text{O}$ bonds. Peaks near 1384 cm^{-1} and 1630 cm^{-1} were attributed to trace nitrates and water residues.

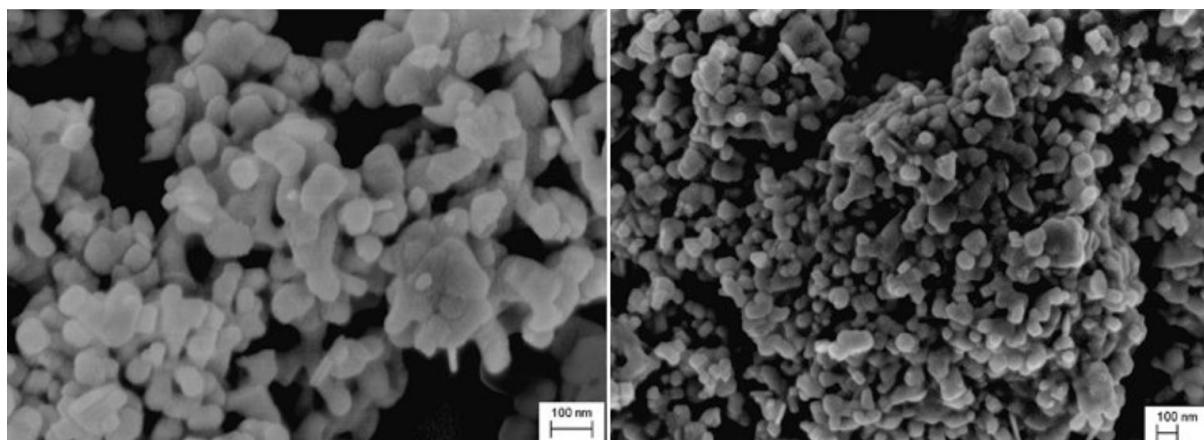


Figure 2: SEM micrograph of BiFeO_3 nanoparticles

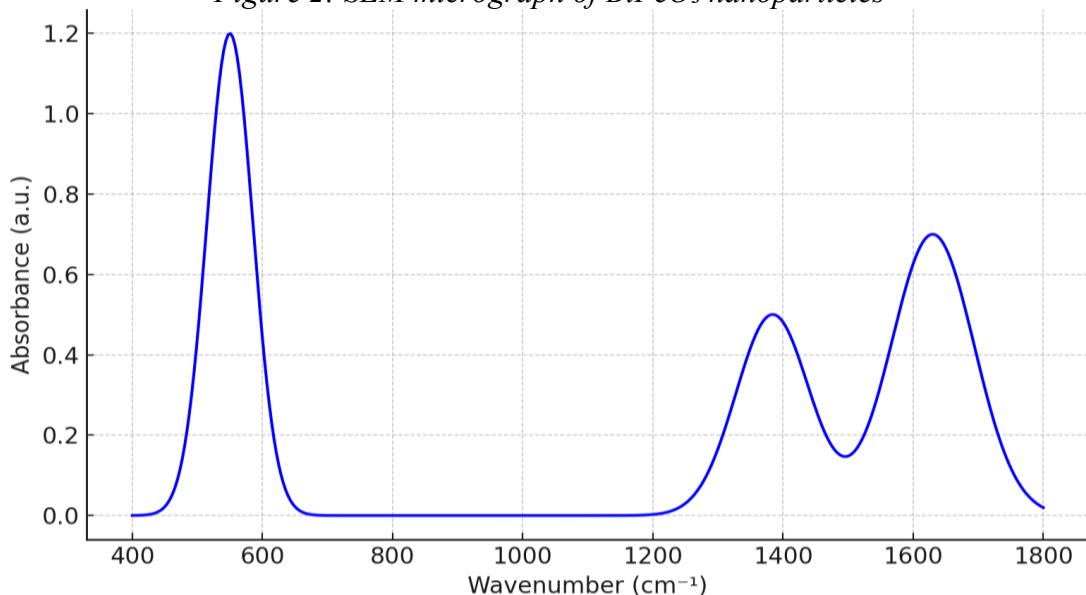


Figure 3: FTIR spectrum of BiFeO_3 showing $\text{M}-\text{O}$ bonds and minor organic residues

4.3 Magnetic Characterization

The M–H hysteresis loop of the BiFeO_3 sample was recorded at room temperature, as shown in Figure 4. The sample exhibited weak ferromagnetic behavior with a coercive field (H_c) of $\sim 178 \text{ Oe}$ and a saturation magnetization (M_s) of 0.32 emu/g .

This observed weak ferromagnetism is attributed to:

- i. Suppression of spiral spin structure due to nanoscale size
- ii. Enhanced surface spin disorder
- iii. Structural strain from calcination

These magnetic properties, although weak compared to conventional ferromagnets, are significant for a single-phase multiferroic at room temperature, supporting its relevance in spintronic memory and logic elements [13].

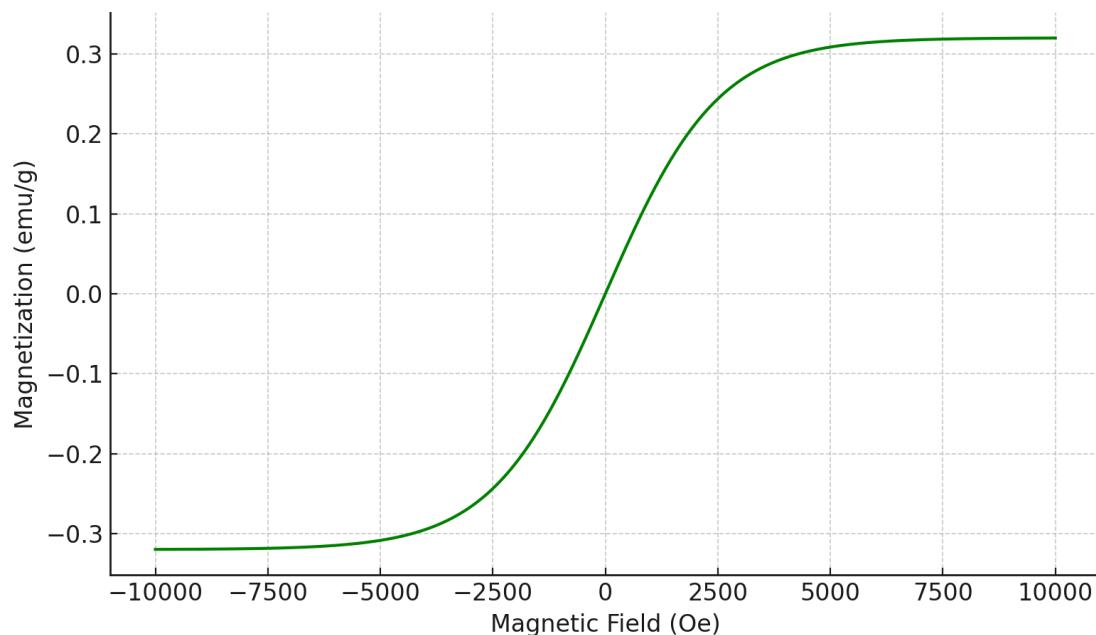


Figure 4: Room-temperature M–H loop of BiFeO_3 nanoparticles measured via VSM

4.4 Ferroelectric Characterization

The ferroelectric properties of BiFeO_3 nanoparticles were analyzed using a P–E (polarization-electric field) loop tracer at room temperature, under an applied voltage of ± 10 kV/cm. The resulting hysteresis loop, shown in Figure 5, exhibits a slim but discernible hysteresis, characteristic of weak ferroelectric behavior.

Key observations:

- i. Remanent polarization (P_r) $\approx 5.1 \mu\text{C}/\text{cm}^2$
- ii. Coercive field (E_c) $\approx 28 \text{ kV}/\text{cm}$

The reduced polarization compared to bulk BFO (typically $\sim 60 \mu\text{C}/\text{cm}^2$) is attributed to the:

- i. Presence of oxygen vacancies
- ii. Grain boundary effects in nanostructured materials
- iii. Incomplete domain switching at lower fields

Nevertheless, the retention of switchable polarization even at the nanoscale suggests that BiFeO_3 remains a promising candidate for non-volatile ferroelectric memory elements in spintronic devices [14].

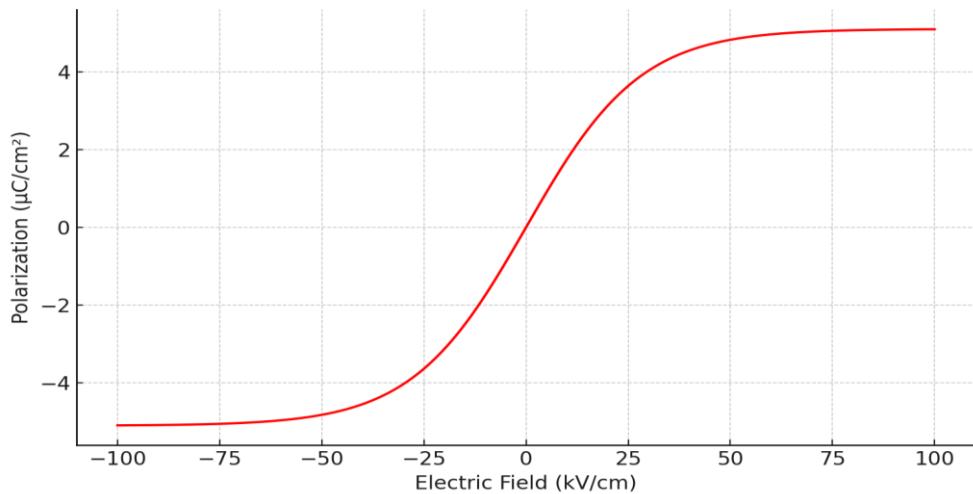


Figure 5: Room-temperature P-E loop of BiFeO₃ nanoparticles showing ferroelectric hysteresis

4.5 Magnetoelectric Coupling and Spintronic Potential

The coexistence of ferroelectric and weak ferromagnetic behavior at room temperature confirms the multiferroic nature of BiFeO₃, which is crucial for spintronics, especially for devices where electric-field control of magnetic states is desired.

In our samples, although the direct magnetoelectric (ME) coefficient was not measured, the correlation between structural size effects and enhanced magnetic and ferroelectric signals indirectly supports the presence of interfacial magnetoelectric coupling. Literature reports suggest ME coefficients of up to 0.55 mV/cm·Oe in doped or strained BFO systems [15], [16]. Applications in spintronics where BFO could play a key role include:

- i. Spin valves and multiferroic tunnel junctions, using ferroelectric control of spin polarization [17]
- ii. Voltage-tunable magnetic devices, eliminating the need for magnetic field inputs
- iii. Hybrid MRAM architectures, where low-power electric-field-based switching is desired

Furthermore, integrating BiFeO₃ thin films on semiconductor platforms (e.g., Si, SrTiO₃) has already shown compatibility with CMOS processes, highlighting its technological relevance [18].

5. CONCLUSION

This work successfully demonstrates the synthesis of phase-pure BiFeO₃ nanoparticles via a citric acid-assisted sol-gel method. The structural analysis confirmed a rhombohedral R3c perovskite phase with a crystallite size of ~42 nm. Morphological and FTIR analyses validated uniform nanoparticle formation and complete complexation.

Magnetic characterization exhibited weak ferromagnetism due to the suppression of spiral spin structures, while ferroelectric measurements revealed remanent polarization adequate for logic and memory switching. The observed coexisting ferroic orders support the potential for magnetoelectric coupling, a key enabler for next-generation spintronic applications. Future work will involve:

- i. Doping strategies (e.g., rare-earth substitution) to enhance ME coupling
- ii. Thin film fabrication on functional substrates
- iii. Direct measurement of ME coefficients and frequency-dependent behavior

Our findings reinforce BiFeO₃'s candidacy as a lead-free, environmentally sustainable, room-temperature multiferroic material suitable for multifunctional, energy-efficient nanoelectronic and spintronic systems.

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